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8 Chemical Equations and Reactions

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Chemistry Chapter 8 Review: Chemical Equations ____1. In a chemical equation, the __1__ are found on the left side of the arrow, ____2. while, the __2__ are found ...

Chemistry Chapter 8 Review: Chemical Equations

Modern Chemistry 1 Chemical Equations and Reactions CHAPTER 8 REVIEW Chemical Equations and Reactions Teacher Notes and Answers Chapter 8 SECTION 1 SHORT ANSWER 1. a. d b. a c. b d. f e. c 2. 8,4,9 3. a. 12 atoms b. 16 atoms c. 51 atoms d. 3 1024 atoms 4. 2Al(s) + 3CuF2(aq) 2AlF3(aq) + 3Cu(s) 5. NaCl(aq) + AgNO3(aq) AgCl(s) + NaNO3(aq) 6. a.

CHAPTER 8 REVIEW Chemical Equations and Reactions

CHAPTER 8 REVIEW Chemical Equations and Reactions MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. b A balanced chemical equation represents all the following except (a) experimentally established facts. (b) the mechanism by which reactants combine to form products.

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Modern Chemistry Chapter 8 Review Chemical Equations And ...

Chemical Equations and Reactions The evolution of energy as light and heat is an indication that a chemical reaction is taking place. CHAPTER 8 Fireworks. CHEMICAL EQUATIONS AND REACTIONS 261 SECTION 1 O BJECTIVES List three observations that suggest that a chemical reaction has taken place.

CHAPTER 8 Chemical Equations and Reactions

Balanced chemical equations show that the elements, atoms and mass on the left equals the elements, atoms and mass on the right: the matter that goes in must come out. Example: The following chemical equation shows that ammonium dichromate breaks down into nitrogen, chromium(III) oxide, and water.

St. Charles Parish Public School System

Katalynn Sutton Chapter 8 Review Chemical Equations 8.1 A chemical equation is a shorthand expression for a chemical change or reaction which uses chemical symbols and formulas of the reactant and products and other symbolic terms to represent a chemical reaction.

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CHAPTER 8 REVIEW Chemical Equations and Reactions SECTION 3 SHORT ANSWER Answer the following questions in the space provided 1 List four metals that will not replace hydrogen in an acid Choose from Cu, Ag, Au, Pt, Sb, Bi, and Hg 2 Consider the metals iron and silver, both listed in Table 3on page 286 of the text Which one Math Tutor Balancing ...

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CHAPTER 8 REVIEW Chemical Equations and Reactions SECTION 2 SHORT ANSWER Answer the following questions in the space provided 1 Match the equation type on the left to its representation on the right c synthesis (a) AX BY → AY BX d decomposition (b) A BX → AX B b single-displacement (c) A B → AX a double-displacement (d) AX → A X

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Chapter 8 is the first chapter of the second semester. Here, we will cover chemical reactions and equations. Students will first learn about the basics of chemical reactions and how to recognize...

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Name Date CHAPTER 8 REVIEW Class 5 E-ci7b IV hemical Equations and Reactions MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. b d A balanced chemical equation represents all the following except (a) experimentally established facts. (b) the mechanism by which reactants combine to form products.

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8. 9. 10. The actual knowledge about what products are produced In a chemical reaction IS obtained by (a) inspecting the chemical equation; (b) balancing the chemical equation; (c) laboratory analysis; (d) BT1ting a word equation. Once the correct formula for a reactant in an equabon has been written, the